

Alkynes as Synthetic Equivalents to Stabilized Wittig Reagents: Intra- and Intermolecular Carbonyl Olefinations Catalyzed by Ag(I), BF₃, and HBF₄

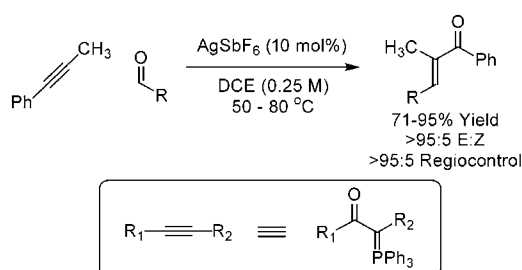
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ABSTRACT



The first use of cationic silver (AgSbF₄) as a catalyst for intra- and intermolecular alkyne–carbonyl coupling to form conjugated enones is described, and a comparison to corresponding Brønsted acid (HBF₄) and Lewis acid (BF₃) catalyst systems is made. Notably, intermolecular coupling proceeds stereoselectively to afford the corresponding trisubstituted enones as single geometrical isomers. This transformation represents a completely atom economical alternative to the use of stabilized Wittig reagents in carbonyl olefination and may be viewed as a formal alkyne–carbonyl metathesis.

Despite tremendous advances in transition metal-catalyzed alkene–alkene, alkene–alkyne, and alkyne–alkyne metathesis,¹ related metal-catalyzed metatheses of carbonyl partners are far less developed.^{2,3} The most broadly utilized method for carbonyl olefination is the Wittig reaction. However, many commonly employed variants of the Wittig reaction suffer due to the stoichiometric production of triphenylphos-

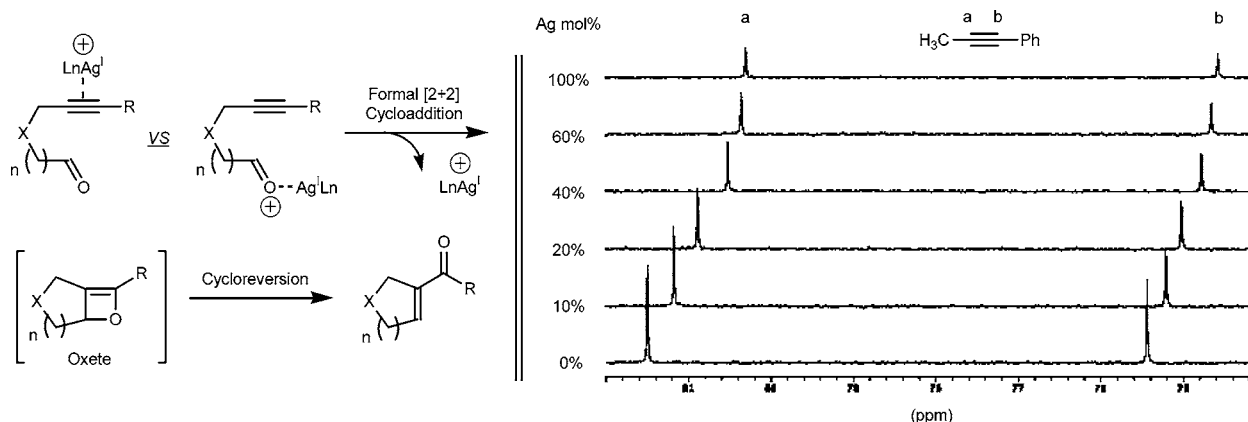
phine oxide. In view of this deficiency, we were inspired by reports of Brønsted and Lewis acid-catalyzed cyclizations of acetylenic ketones to afford conjugated enones: a formal alkyne–carbonyl metathesis.^{4–6} In this account, we report the first late transition metal catalyst for formal alkyne–

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(2) For selected reviews encompassing carbonyl olefinations that are postulated to occur through the intermediacy of metal carbene complexes, see: (a) Nenajdenko, V. G.; Korotchenko, V. N.; Shastin, A. V.; Balenkova, E. S. *Russ. Chem. Bull., Int. Ed.* **2004**, *53*, 1034. (b) Kuhn, F. E.; Santos, A. M. *Mini-Rev. Org. Chem.* **2004**, *1*, 55. (c) Breit, B. *Angew. Chem., Int. Ed.* **1998**, *37*, 453. (d) Ephritikhine, M. *Chem. Commun.* **1998**, 2549. (e) McMurry, J. E. *Chem. Rev.* **1989**, *89*, 1513.

(3) For selected examples of intramolecular alkylidene-mediated alkene–carbonyl metathesis, see: (a) Fu, G. C.; Grubbs, R. H. *J. Am. Chem. Soc.* **1993**, *115*, 3800. (b) Stille, J. R.; Grubbs, R. H. *J. Am. Chem. Soc.* **1986**, *108*, 3800. (c) Nicalaou, K. C.; Posteema, M. H. D.; Claiborne, C. F. *J. Am. Chem. Soc.* **1996**, *118*, 1565.

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Scheme 1^a

^a Left: Alkyne-carbonyl metathesis via alkyne or aldehyde complexation-initiated oxete formation. Right: ¹³C NMR spectroscopic analysis of an equimolar mixture of 1-phenyl propyne and isobutyraldehyde reveals a substantial upfield shift of the alkyne carbon signals upon addition of AgSbF₆, while signals corresponding to isobutyraldehyde exhibit negligible change.

carbonyl metathesis. Specifically, upon exposure to cationic Ag(I) salts, alkynes and aldehydes undergo intra- and intermolecular alkyne-carbonyl coupling to provide trisubstituted enones. Notably, in the case of intermolecular coupling, complete levels of regio- and stereocontrol are observed. Additionally, as part of a broad effort to develop the use of alkynes as atom economical alternatives to Wittig-type reagents, comparisons are made with related Brønsted acid (HBF₄)- and Lewis acid (BF₃·OEt₂)-catalyzed processes.

Brønsted and Lewis acid-catalyzed couplings of acetylenic carbonyl compounds are likely initiated through activation of the carbonyl partner by complexation of oxygen non-bonding electrons. While transannular ring closures of this type are unlikely to proceed through the intermediacy of *anti*-Bredt oxetes,^{4k,l} simple nontransannular cyclizations may involve a mechanism involving stepwise oxete formation followed by cycloreversion to provide the conjugated enone. Such a mechanism is consistent with isotopic labeling studies on the acid-catalyzed rearrangement of 6-octyn-2-one in the presence of H₂¹⁸O, which occurs without ¹⁸O-incorporation.^{4f} The independent preparation of oxetes, typically under photochemical conditions, and their authenticated cycloreversion lends additional support to this mechanism.⁷

An alternative catalytic mechanism potentially promoted through the use of a “carbophilic” Lewis acid involves alkyne complexation-initiated oxete formation (Scheme 1, right). Predicated on the basis of the well-established ability

of silver(I) salts to form strong π -complexes with alkene and alkyne partners,^{8,9} the intramolecular silver(I)-catalyzed metathesis of acetylenic aldehyde **3a** was explored. Gratifyingly, exposure of **3a** to substoichiometric quantities of AgSbF₆ (10 mol %) in dichloroethane at ambient temperature led to a nearly quantitative isolated yield of the trisubstituted enone **3b**. Withstanding changes in temperature, these conditions proved to be applicable across a diverse set of acetylenic aldehydes **1a–9a**, enabling formation of both five- and six-membered ring products. Additionally, as demonstrated by the cyclization of **10a**, acetylenic ketones in the form of 1,3-diones also participate in the reaction. In each case, Lewis (BF₃·OEt₂)- and Brønsted acid (HBF₄)-catalyzed reactions also were explored. For certain substrates, the silver(I) catalyst is more effective (**5a**, **6a**, **7a**, **9a**, **10a**), while in other cases (**1a**, **2a**) use of the Lewis or Brønsted acid catalyst is preferred (Table 1).

Intermolecular carbonyl metathesis of aliphatic and aromatic aldehydes (300 mol %) with 1-phenyl-1-propyne (100 mol %) affords enones **11b–14b**. Complete regioselection is observed. Additionally, the trisubstituted alkenes appear as single geometrical isomers. Under a range of related conditions, terminal alkynes such as phenylacetylene provide dramatically reduced yields of the analogous 1,2-disubstituted enone products (Table 2).

Addition of AgSbF₆ to an equimolar solution of 1-phenyl-1-propyne and isobutyraldehyde results in significant upfield

(5) Intermolecular alkyne-aldehyde metathesis has been achieved using stoichiometric Lewis acid promoters: (a) Viswanathan, G. S.; Li, C.-J. *Tetrahedron Lett.* **2002**, 43, 1613. (b) Hayashi, A.; Yamaguchi, M.; Hirama, M. *Synlett* **1995**, 195.

(6) A single example of catalytic intermolecular alkyne-aldehyde metathesis has been described. This Yb(OTf)₃-catalyzed process enables the formation of chalcones from aromatic alkynes: Curini, M.; Epifano, F.; Maltese, F.; Rosati, O. *Synlett* **2003**, 552.

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(9) For selected examples of Ag(I)-alkyne complexes characterized by single-crystal X-ray diffraction analysis, see: (a) Ferrara, J. D.; Djebli, A.; Tessier-Youngs, C.; Youngs, W. J. *J. Am. Chem. Soc.* **1988**, 110, 647. (b) Gleiter, R.; Karcher, M.; Kratz, D.; Ziegler, M. L.; Nuber, B. *Chem. Ber.* **1990**, 123, 1461. (c) Meier, H.; Dai, Y. *Tetrahedron Lett.* **1993**, 34, 5277. (d) Nishinaga, T.; Kawamura, T.; Komatsu, K. *Chem. Commun.* **1998**, 2263. (e) Schulte, P.; Behrens, U. *J. Organomet. Chem.* **1998**, 563, 235. (f) Chi, K.-M.; Lin, C.-T.; Peng, S.-M.; Lee, G.-H. *Organometallics* **1996**, 15, 2660.

Table 1. Intramolecular Alkyne–Carbonyl Coupling Catalyzed by Ag(I)SbF₆, HBF₄, and BF₃•OEt₂^a

conditions A AgSbF ₆ (10 mol%)		conditions B BF ₃ (OEt ₂) (20 mol%)		conditions C HBF ₄ (20 mol%)	
substrate	product	substrate	product	substrate	product
 1a	 1b { <ul style="list-style-type: none"> A: 60%, 80 °C B: 81%, 50 °C C: 61%, 50 °C 	 6a	 6b { <ul style="list-style-type: none"> A: 81%, 80 °C B: 70%, 50 °C C: 63%, 50 °C 		
 2a	 2b { <ul style="list-style-type: none"> A: 54%, 25 °C B: 71%, 25 °C C: 71%, 25 °C 	 7a E = CO ₂ C ₂ H ₅	 7b { <ul style="list-style-type: none"> A: 74%, 25 °C B: 48%, 25 °C C: 58%, 25 °C 		
 3a	 3b { <ul style="list-style-type: none"> A: 99%, 25 °C B: 89%, 25 °C C: 94%, 25 °C 	 8a	 8b { <ul style="list-style-type: none"> A: 93%, 25 °C B: 96%, 25 °C C: 93%, 25 °C 		
 4a	 4b { <ul style="list-style-type: none"> A: 68%, 25 °C B: 48%, 25 °C C: 63%, 25 °C 	 9a	 9b { <ul style="list-style-type: none"> A: 93%, 80 °C B: 66%, 50 °C C: 61%, 50 °C 		
 5a	 5b { <ul style="list-style-type: none"> A: 82%, 25 °C B: 31%, 25 °C C: 22%, 25 °C 	 10a	 10b { <ul style="list-style-type: none"> A: 89%, 25 °C B: 56%, 25 °C C: 59%, 25 °C 		

^a See Supporting Information for detailed experimental procedures.

shifts of the alkyne signals in the ¹³C NMR spectra. The aldehyde chemical shifts remain essentially unchanged. While these data suggest alkyne-initiated oxete formation, a Curtin–Hammett scenario in which coupling occurs through the intermediacy of small quantities of silver–aldehyde complex cannot be discounted. While it is possible that AgSbF₆ may react with residual moisture to produce

Table 2. Intermolecular Alkyne–Carbonyl Coupling Catalyzed by Ag(I)SbF₆, HBF₄, and BF₃•OEt₂^a

Conditions A, B, or C
DCE (0.25 M), T °C

11b

{	A: 71%, 80 °C
{	B: 73%, 80 °C
{	C: 73%, 80 °C

12b

{	A: 74%, 80 °C
{	B: 64%, 80 °C
{	C: 53%, 80 °C

13b

{	A: 95%, 50 °C
{	B: 44%, 50 °C
{	C: 48%, 50 °C

14b

{	A: 80%, 50 °C
{	B: 63%, 80 °C
{	C: 59%, 80 °C

^a See Supporting Information for detailed experimental procedures.

Brønsted acids that may catalyze enone formation, the olefination proceeds readily in the presence of homogeneous and heterogeneous bases such as *i*-Pr₂NEt, 2,6-di-*tert*-butylpyridine, and K₂CO₃ (Scheme 1, left).

In summary, AgSbF₆ catalyzes intra- and intermolecular alkyne–carbonyl coupling to form conjugated enones. A comparison to corresponding Brønsted acid (HBF₄) and Lewis acid (BF₃) catalyst systems reveals that the AgSbF₆-catalyzed process is moderately more efficient in certain cases. Notably, intermolecular coupling proceeds stereoselectively to afford the corresponding trisubstituted enones as single geometrical isomers. This transformation represents a completely atom economical alternative to the use of stabilized Wittig reagents in carbonyl olefination and may be viewed as a formal alkyne–carbonyl metathesis. Future studies will focus on the development of improved second-generation catalyst systems.

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Supporting Information Available: Spectral data for all new compounds (¹H NMR, ¹³C NMR, IR, HRMS). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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